

## EXOGENOUS APPLICATION OF SODIUM NITROPRUSSIDE ALLEVIATED CADMIUM INDUCED CHLOROSIS, PHOTOSYNTHESIS INHIBITION AND OXIDATIVE STRESS IN CUCUMBER

LIXU YU, RONGXIA GAO, QINGHUA SHI\*, XIUFENG WANG, MIN WEI AND FENGJUAN YANG

State Key Laboratory of Crop Biology, College of Horticulture Science and Engineering,  
Shandong Agricultural University, Taian 271018, P.R. China

\*Corresponding author's e-mail: sqhjc@163.com, qhshi@sdau.edu.cn; Tel: 86-538-8242201

### Abstract

Cadmium (Cd) is one heavy metal which is toxic and can cause physiological disorder of plants. In the present experiment, effects of Sodium nitroprusside (SNP), a NO donor, on the phenotype, photosynthesis and oxidative stress of cucumber leaves were studied under Cd stress. Application of 100 $\mu$ M CdCl<sub>2</sub> induced obvious chlorosis of cucumber leaves with lower pigment content and lower net photosynthetic rate (P<sub>N</sub>). Higher accumulations of H<sub>2</sub>O<sub>2</sub> and thiobarbituric acid-reactive substances (TBARS) were observed in Cd-treated cucumber leaves, which indicated Cd induced oxidative stress to cucumber leaves. Application of 100 $\mu$ M SNP reversed the chlorosis and lower P<sub>N</sub> induced by Cd treatment. Activities of reactive oxygen species (ROS) scavenging enzymes and antioxidant capacity expressed by  $\alpha,\alpha$ -diphenyl- $\beta$ -picrylhydrazyl (DPPH) scavenging activity, chelating activity of ferrous ions and hydroxyl radical (OH $\cdot$ ) scavenging activity were significantly increased by SNP under Cd stress. On the contrary, the above effects of SNP were not observed after the application of potassium ferrocyanide which is an analog of SNP that does not release NO. Therefore, it could be concluded that the NO from SNP might account for the alleviating effect of Cd stress on cucumber plants.

### Introduction

Cadmium (Cd) is one of widespread heavy metals; it can be easily up-taken by plants and enter the food chain to be particularly dangerous for human health. The toxic levels of Cd may originate from soil with characteristics of abundant Cd or agricultural manufacturing, mining and other waste disposal practices, or from the application of Cd-containing pesticides and phosphate fertilizers (Radotic *et al.*, 2000). Some investigations indicated that Cd led to deficiency of iron and induced the inhibition of chlorophyll biosynthesis which resulted in a decline of the photosynthetic rate (Wahid *et al.*, 2008; Zulfqar *et al.*, 2012). Like other stress, Cd toxicity also destroys the balance between reactive oxygen species (ROS) and its scavenger, and induces oxidative stress to plants (Markovska *et al.*, 2009).

Nitric oxide (NO) is a small, ubiquitous, highly diffusible gaseous bioactive molecule. Its chemical properties make NO a versatile signal molecule that functions through interactions with cellular targets via either redox or additive chemistry (Lamattina *et al.*, 2003). Recent studies provided increasing evidences that NO is involved in modulating plants tolerance to abiotic stress including salt stress, temperature stress, water stress and heavy metal toxicity (Mahmood *et al.*, 2009; Siddiqui *et al.*, 2011). Cucumber (*Cucumis sativus* L.) is one of the important vegetables in the world, and it is sensitive to Cd toxicity (Feng *et al.*, 2010). It has been reported that NO was involved in cucumber tolerance to drought and salt stress (Arasimowicz-Jelonek & Foryszak-Wieczorek, 2009; Shi *et al.*, 2007), however, to our knowledge, there are few investigations about NO regulating Cd toxicity in cucumber. In the present study, possible mechanism of exogenous SNP alleviating Cd toxicity in cucumber was studied by investigating photosynthetic parameters and antioxidant metabolism, especially activities of antioxidant enzymes including superoxide dismutase (SOD), catalase (CAT), guaiacol peroxidase (GPX), ascorbate peroxidase (APX), dehydroascorbate reductase (DHAR) and

glutathione reductase (GR) which are involved in the ROS scavenging.

### Material and Methods

Cucumber (*Cucumis sativus* L. cv. Jinyou 4) seeds (Provided by Tianjin Cucumber Research Institute) were germinated on moisture filter paper in the dark at 28°C for 2 days, and germinated seedlings were transferred to the growth chamber filled with vermiculite and grown in greenhouse for 11 days. Then they were transplanted into 5L black plastic containers containing aerated full nutrient solution: 4 mM Ca(NO<sub>3</sub>)<sub>2</sub>, 4 mM KNO<sub>3</sub>, 2.5 mM KH<sub>2</sub>PO<sub>4</sub>, 2 mM MgSO<sub>4</sub>, 29.6  $\mu$ M H<sub>3</sub>BO<sub>3</sub>, 10  $\mu$ M MnSO<sub>4</sub>, 50  $\mu$ M Fe-EDTA, 1.0  $\mu$ M ZnSO<sub>4</sub>, 0.05  $\mu$ M H<sub>2</sub>MoO<sub>4</sub>, 0.95  $\mu$ M CuSO<sub>4</sub>, with 3 seedlings per container. After 10 days of pre-culture, the treatments were started. The experimental design consisted of a control and three treatments (Cd: 100  $\mu$ M CdCl<sub>2</sub> treatment; Cd+ SNP: 100  $\mu$ M CdCl<sub>2</sub> +100  $\mu$ M SNP treatment; Cd+SF, 100  $\mu$ M CdCl<sub>2</sub> +100  $\mu$ M potassium ferrocyanide treatment) and was arranged in a randomized, complete block design with three replicates. Single container was used as treatment unit. The plants were cultivated in a greenhouse where the day/night time and temperature were about 12h/12h and 23-28 /18-23 , following 12 days of treatment application, samples were collected for physiological parameters.

### Determination of plant growth and Fe concentration:

After 12 days of treatments, the cucumber plants were harvested, divided in to shoots and roots. The samples were dried at 70°C to constant weight and weighted. For determination of Fe concentrations, 0.3g mixed powders of dried leaves were transferred to Kjeldahl flask, in which 5mL 30% H<sub>2</sub>O<sub>2</sub> and 10mL of concentrated H<sub>2</sub>SO<sub>4</sub> solution were added, and the sample was digested until the solution become clear. Then it was transferred to the volumetric flask and volume up to 100 ml was made up with the twice-distilled water. The concentration of iron was determined by flame atomic absorption spectrometry (Koniecznyński & Wesolowski, 2007).

**Determination of H<sub>2</sub>O<sub>2</sub> content:** H<sub>2</sub>O<sub>2</sub> content was determined according to the method reported by Patterson *et al.*, (1984). The assay was based on the absorbance change of the titaniumperoxide complex at 415nm. Absorbance values were quantified using standard curve generated from known concentrations of H<sub>2</sub>O<sub>2</sub>.

**Determination of lipid peroxidation:** Lipid peroxidation (LPO) was estimated by measuring the concentration of thiobarbituric acid reactive substances (TBARS) using the thiobarbituric acid method described by Heath *et al.*, (1968). 0.3g of leaves were ground with 3ml of 0.1% (w/v) trichloroacetic acid (TCA). The homogenate was centrifuged at 10,000 × *g* for 10 min and 3 ml of 20% TCA containing 0.5% (w/v) 2-thiobarbituric acid (TBA) was added to 1ml of supernatant. The mixture was heated at 95°C for 30 min and the reaction was stopped by quickly placing in an ice-bath. The cooled mixture was centrifuged at 10,000 × *g* for 10 min, and the absorbance of the supernatant was recorded at 532 and 600 nm. After subtracting the non-specific absorbance at 600 nm, the TBARS concentration was determined by its extinction coefficient of 155 mM<sup>-1</sup> cm<sup>-1</sup>.

**Enzyme extraction:** For enzyme assays, 0.3g leaves were ground with 3 ml ice-cold 25 mM HEPES buffer (pH 7.8) containing 0.2 mM EDTA, 2 mM ascorbate and 2% PVP. The homogenates were centrifuged at 4°C for 20 min at 12,000 × *g* and the resulting supernatants were used for determination of enzymatic activities (Zhu *et al.*, 2004). All spectrophotometric analyses were conducted on a SHIMADZU UV-2450PC spectrophotometer.

**Determination of antioxidant enzyme activities:** SOD activity was assayed by measuring its ability to inhibit the photochemical reduction of nitroblue tetrazolium following the method of Stewart & Bewley (1980). CAT activity was determined as the decline in absorbance at 240 nm due to the decrease of extinction of H<sub>2</sub>O<sub>2</sub> using the method of Patra *et al.*, (1978). GPX activity was determined as the increase in absorbance at 470 nm due to guaiacol oxidation (Nickel *et al.*, 1969). APX activity was determined by the decrease in absorbance at 290 nm as ascorbate was oxidized (Nakano *et al.*, 1981). DHAR activity was assayed by measuring the increase in absorbance at 265nm due to reduced ascorbate formation (Nakano *et al.*, 1981). GR activity was measured according to Foyer *et al.*, (1976), which depended on the rate of decrease in the absorbance of NADPH at 340 nm.

**Determination of photosynthetic parameters:** The content of photosynthetic pigments was measured according to the method of Arnon (1949). Net photosynthetic rate was measured by a portable photosynthesis system Li-6400 (Li-Cor, Lincoln, USA) at ambient CO<sub>2</sub> concentration of 340 μmol mol<sup>-1</sup> and a photon flux density of 1000 μmol m<sup>-2</sup> S<sup>-1</sup>. Fv/Fm was measured with a pulse amplitude modulated system (model FMS2, Hansatech Instrument, UK) according to Burzyński & Klobus (2004).

**Determination of glutathione content:** frozen sample (0.3g) was ground with a mortar and pestle in 3ml of 0.5mM EDTA solution containing 3% TCA at 4°C. The homogenate was centrifuged at 15,000 × *g* for 10 min. Then 0.2mL supernatant was added to 1.5mL of 50mM potassium phosphate buffer (pH 7.0) containing 0.2mM 5,5'-dithio-bis (2-nitrobenzoic) (DTNB). The mixture was incubated at 30°C for 2 min. Absorbance was determined at 412nm and the glutathione (GSH) concentration was calculated by comparison to a standard curve (Ellman, 1959).

**Determination of antioxidant activity:** For determination of antioxidant activities indicated as DPPH scavenging capacity, Hydroxyl radical (OH) scavenging capacity and Fe<sup>2+</sup>-chelating capacity, 0.3g sample was suspended in 3ml of serine borate buffer (100mM Tris-HCl, 10mM borate, 5mM serine, and 1 mM diethylenetriaminepentacetic acid, pH 7.0). The slurry was centrifuged at 5000 × *g* for 10 min at 4°C and the supernatants were used for the in vitro antioxidant assays. All samples were placed on ice during the experiments. The analysis of antioxidant capacity was done according to the method of Manda *et al.*, (2010).

**The statistics method:** Values presented were means ± 1 standard deviation (SD) of three replicates. Statistical analyses were carried out by analysis of variance (ANOVA) using SPSS10.0 software. Differences between treatments were analyzed by the Duncan's multiple range test.

## Results

**Effects of exogenous SNP on leaf symptom, photosynthetic parameters and plant growth of cucumber under Cd stress:** Cd stress dramatically induced cucumber leaf chlorosis, in accordance with the chlorosis, the contents of chlorophyll a, chlorophyll b and carotenoid were significantly decreased by Cd treatment (*p*<0.05) (Fig. 1 & Table 1). Application of SNP reversed the chlorosis of cucumber leaves induced by Cd stress, while application of potassium ferrocyanide, an analog of SNP that does not release NO, did not show significant effect on Cd-induced cucumber leaf chlorosis and lower pigment contents. Cd stress significantly inhibited the net photosynthetic rate (P<sub>N</sub>) (*p*<0.05) and Fv/Fm ratio (*p*<0.05) and reduced stomatal conductance (G<sub>s</sub>) (*p*<0.05), which were accompanied by higher intracellular CO<sub>2</sub> concentration (C<sub>i</sub>) levels. SNP application significantly increased G<sub>s</sub> and alleviated the inhibition of P<sub>N</sub> and Fv/Fm by Cd treatment, and resulted in lower C<sub>i</sub>. These effects of SNP on photosynthetic parameters were not obtained by application of potassium ferrocyanide. Consistent with the effects on pigments and photosynthesis, Cd stress significantly inhibited the growth of cucumber shoots and roots, and SNP application reduced the inhibition, while potassium ferrocyanide did not show obvious influence on cucumber plant growth under Cd stress (Fig. 3).

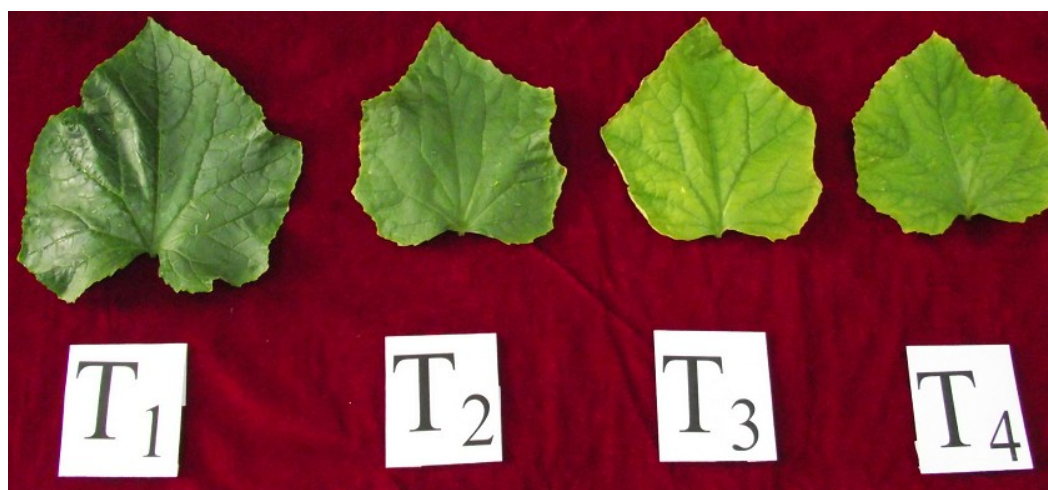


Fig. 1. Effects of exogenous sodium nitroprusside on the phenotype of Cd-treated cucumber leaves. T1, Control; T2, 100 $\mu$ M CdCl<sub>2</sub> + 100 $\mu$ M SNP; T3, 100 $\mu$ M CdCl<sub>2</sub>; T4, 100 $\mu$ M CdCl<sub>2</sub> + 100  $\mu$ M potassium ferrocyanide.

**Table 1. Effects of exogenous sodium nitroprusside on pigment contents, P<sub>N</sub> (net photosynthetic rate), E (transpiration rate), Ci (intracellular CO<sub>2</sub> concentration), g<sub>s</sub> (stomatal conductance) and Fv/Fm in Cd-treated cucumber leaves.**

Treatment	Parameters							
	Chl a	Chl b	Carotenoid	P <sub>N</sub>	E	g <sub>s</sub>	Ci	Fv/Fm
CK	1.41±0.05a	0.26±0.04a	0.36±0.02a	17.57±0.75a	10.15±0.67a	0.42±0.04a	295.0±7.00c	0.84±0.007a
Cd	0.68±0.07c	0.16±0.02c	0.18±0.01c	7.87±0.50c	2.76±0.44c	0.09±0.01c	353.7±4.93a	0.75±0.011c
SNP	0.94±0.07b	0.19±0.02b	0.29±0.04b	12.13±0.70b	6.48±0.79b	0.29±0.05b	318.7±2.89b	0.80±0.011b
SF	0.70±0.06c	0.15±0.03c	0.19±0.02c	8.2±1.06c	2.83±0.62c	0.10±0.02c	345.3±11.93a	0.76±0.013c

Note: Data are means ± SD of three replicates. Mean values followed by different letters (a-c) are significantly different ( $p < 0.05$ ). CK, Control; Cd, 100  $\mu$ M CdCl<sub>2</sub>; SNP, 100  $\mu$ M CdCl<sub>2</sub> +  $\mu$ M SNP; SF, 100  $\mu$ M CdCl<sub>2</sub> + 100  $\mu$ M potassium ferrocyanide. Units of Chl a, Chl b and carotenoid were mg g<sup>-1</sup> FW. Units of P<sub>N</sub>, E, g<sub>s</sub> and Ci were  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>, mmol m<sup>-2</sup> s<sup>-1</sup>, mol m<sup>-2</sup> s<sup>-1</sup> and  $\mu$ L L<sup>-1</sup>, respectively

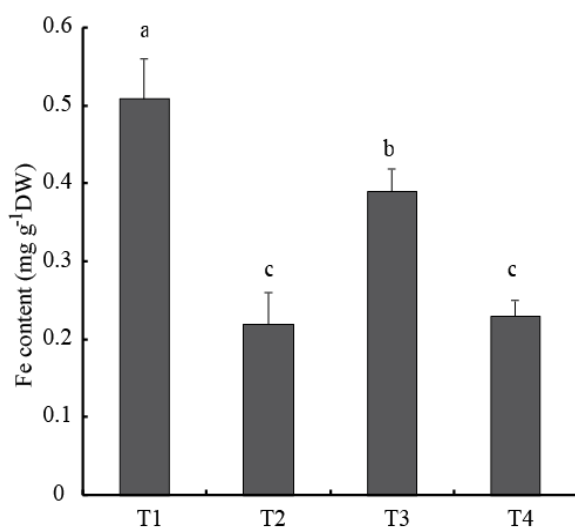


Fig. 2. Effects of exogenous sodium nitroprusside on the Fe content in Cd-treated cucumber leaves. Data are means ± SD of three replicates. Mean values followed by different letters (a-c) are significantly different ( $p < 0.05$ ). T1, Control; T2, 100 $\mu$ M CdCl<sub>2</sub>; T3, 100 $\mu$ M CdCl<sub>2</sub> + 100 $\mu$ M SNP; T4, 100 $\mu$ M CdCl<sub>2</sub> + 100 $\mu$ M potassium ferrocyanide.

**Effects of exogenous SNP on Fe content in cucumber leaves under Cd stress:** Compared to control, Cd treatment significantly reduced Fe accumulation in cucumber leaves ( $p < 0.05$ ), and application of SNP significantly increased Fe content of cucumber leaves ( $p < 0.05$ ), while no similar results were observed by application of potassium ferrocyanide under Cd stress (Fig. 2).

**Effects of exogenous SNP on H<sub>2</sub>O<sub>2</sub> and lipid peroxidation of cucumber leaves under Cd stress:** Compared to the control, Cd treatment dramatically induced accumulation of H<sub>2</sub>O<sub>2</sub> ( $p < 0.05$ ), in accordance with higher H<sub>2</sub>O<sub>2</sub> concentration, TBARS, as an indicator of lipid peroxidation level, were significantly increased in cucumber leaves under Cd stress ( $p < 0.05$ ). The excessive H<sub>2</sub>O<sub>2</sub> accumulation of cucumber leaves induced by Cd were dramatically scavenged by application of SNP ( $p < 0.05$ ), which also resulted in lower lipid peroxidation. The effects of SNP on H<sub>2</sub>O<sub>2</sub> and lipid peroxidation could not be gotten by application of potassium ferrocyanide (Fig.4).

**Effects of exogenous SNP on antioxidant system in cucumber leaves under Cd stress:** As shown in Table 2, Cd stress significantly inhibited the activities of

SOD, CAT, GPX, APX, DR and GR in cucumber leaves ( $p < 0.05$ ). Application of exogenous SNP increased activities of all antioxidant enzymes under Cd stress, while application of potassium ferrocyanide did not significantly affected activities of these antioxidant enzymes.

DPPH scavenging capacity,  $\cdot\text{OH}$  scavenging capacity and metal chelating capacity were usually used to express antioxidant capacity of plant tissues. Compared to the control, Cd stress dramatically decreased antioxidant capacity of cucumber leaves, and reduced DPPH scavenging capacity,  $\cdot\text{OH}$  scavenging capacity and metal chelating capacity by 68%, 35%, 43%, respectively. While application of exogenous SNP significantly alleviated the inhibition of antioxidant capacity by Cd stress, and increased DPPH

scavenging capacity,  $\cdot\text{OH}$  scavenging capacity and metal chelating capacity by 100, 33 and 61% compared to the Cd stress, respectively. The effects of exogenous SNP increasing antioxidant capacity under Cd stress were not found by application of potassium ferrocyanide (Table 2).

Glutathione (GSH), as an important antioxidant, was significantly reduced in cucumber leaves under Cd stress, in which GSH content was only 56% of control ( $p < 0.05$ ). Application of SNP led to significant elevation of GSH content under Cd stress, and 63% higher than Cd treatment was observed ( $p < 0.05$ ). Like other parameters, application of potassium ferrocyanide did not significantly GSH content in cucumber leaves under Cd stress, too (Table 2).

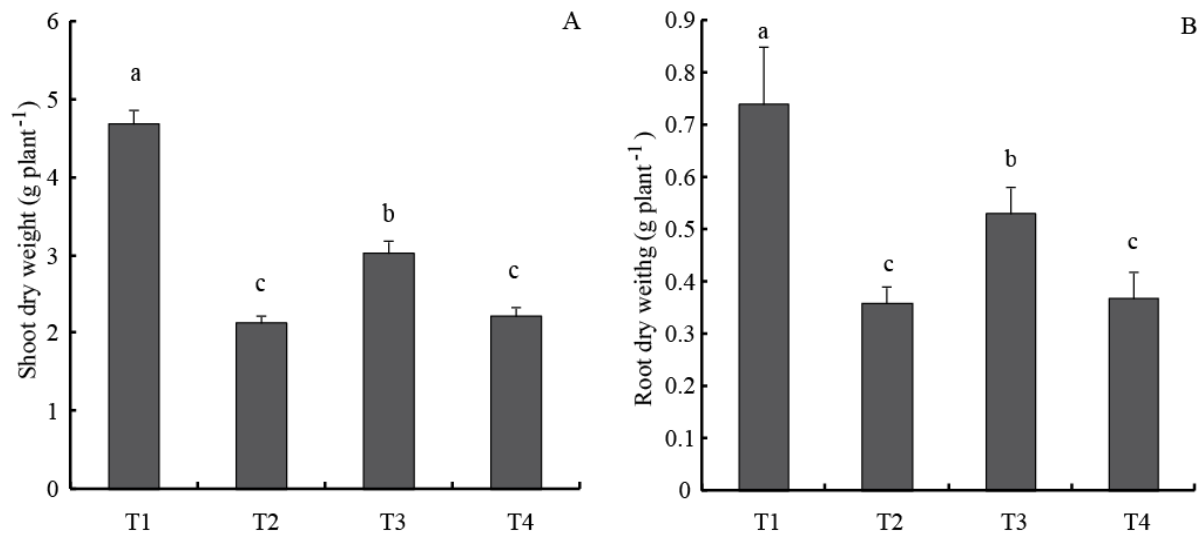


Fig. 3. Effects of exogenous sodium nitroprusside on the dry weight of shoot (A) and root (B) in Cd-treated cucumber plants. Data are means  $\pm$  SD of three replicates. T1, Control; T2, 100  $\mu\text{M}$  CdCl<sub>2</sub>; T3, 100  $\mu\text{M}$  CdCl<sub>2</sub> + 100  $\mu\text{M}$  SNP; T4, 100  $\mu\text{M}$  CdCl<sub>2</sub> + 100  $\mu\text{M}$  potassium ferrocyanide.

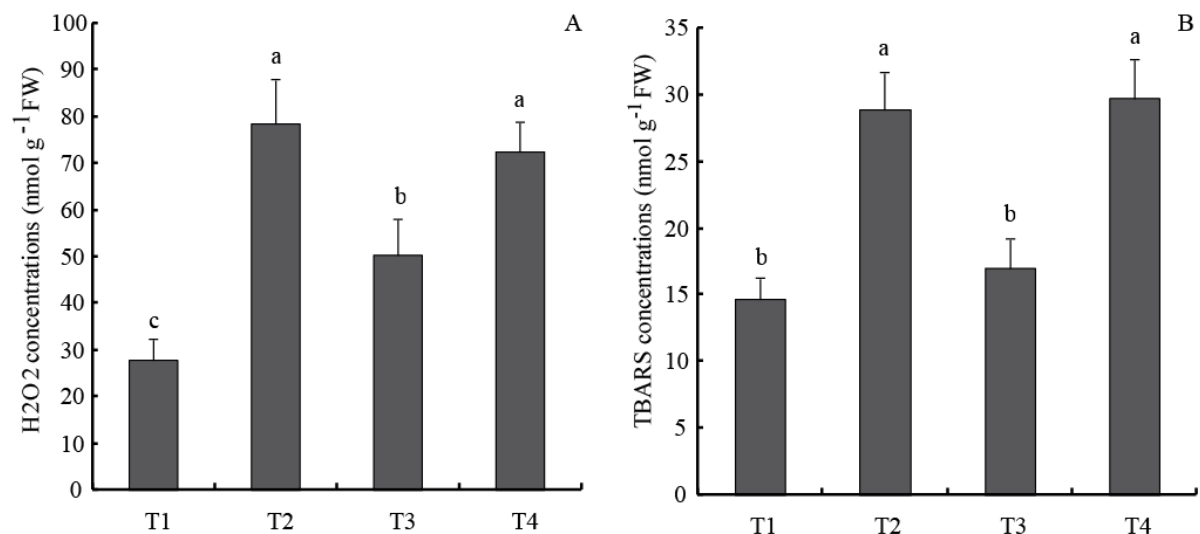


Fig. 4. Effects of exogenous sodium nitroprusside on concentrations of H<sub>2</sub>O<sub>2</sub> (A) and TBARS (B) in Cd-treated cucumber leaves. Data are means  $\pm$  SD of three replicates. Mean values followed by different letters (a-c) are significantly different ( $p < 0.05$ ). T1, Control; T2, 100  $\mu\text{M}$  CdCl<sub>2</sub>; T3, 100  $\mu\text{M}$  CdCl<sub>2</sub> + 100  $\mu\text{M}$  SNP; T4, 100  $\mu\text{M}$  CdCl<sub>2</sub> + 100  $\mu\text{M}$  potassium ferrocyanide.

**Table 2.** Effects of exogenous sodium nitroprusside on activities of SOD, CAT, GPX, APX, DR and GR, DPPH scavenging capacity, ·OH scavenging capacity, metal chelating capacity and GSH contents in Cd-treated cucumber leaves.

Treatment	Parameters									
	SOD activity	CAT activity	GPX activity	APX activity	DR activity	GR activity	DPPH scavenging capacity	·OH scavenging capacity	Metal chelating capacity	GSH content
CK	38.29±3.06a	5.69±0.53a	21.00±3.14a	7.32±0.61a	0.88±0.12a	0.54±0.04a	29.26±3.4a	45.64±2.98a	36.10±2.49a	10.53±1.15a
Cd	15.60±3.18c	2.27±0.14c	8.83±1.59c	3.16±0.29c	0.18±0.02c	0.09±0.02c	9.33±1.45c	29.56±2.49c	20.71±1.90b	5.86±0.82b
SNP	27.29±2.17b	3.61±0.42b	15.84±1.30b	4.83±0.29b	0.35±0.03b	0.26±0.03b	18.65±1.84b	39.02±3.93b	33.41±2.66a	9.57±0.54a
SF	16.62±1.46c	2.37±0.27c	10.01±0.67c	3.32±0.31c	0.20±0.04c	0.11±0.01c	10.65±1.76c	28.70±3.37c	17.47±2.13b	5.17±0.77b

**Note:** Data are means ± SD of three replicates. Mean values followed by different letters (a-c) are significantly different ( $p < 0.05$ ). CK, Control; Cd, 100  $\mu\text{M}$  CdCl<sub>2</sub>; SNP, 100  $\mu\text{M}$  CdCl<sub>2</sub> + 100  $\mu\text{M}$  SNP; SF, 100  $\mu\text{M}$  CdCl<sub>2</sub> + 100  $\mu\text{M}$  potassium ferrocyanide. Units of SOD, CAT, GPX, APX, DR and GR activities were U mg<sup>-1</sup> protein,  $\mu\text{mol H}_2\text{O}_2$  mg<sup>-1</sup> protein min<sup>-1</sup>,  $\mu\text{mol guaiacol mg}^{-1}$  protein min<sup>-1</sup>,  $\mu\text{mol AsA mg}^{-1}$  protein min<sup>-1</sup>,  $\mu\text{mol DHA mg}^{-1}$  protein min<sup>-1</sup> and  $\mu\text{mol NADPH mg}^{-1}$  protein min<sup>-1</sup>, respectively. DPPH scavenging capacity, ·OH scavenging capacity and metal chelating capacity were expressed by percentage (%). The unit of GSH content was nmol g<sup>-1</sup>FW

## Discussion

Cadmium is one of the most common heavy metal pollutant in worldwide soil, and greatly limits crop productivity because of its easy accumulation within plant tissues and its interference with essential physiological processes, among which Fe deficiency induced by Cd was considered to be mainly responsible for chlorosis (Gao *et al.*, 2011). In the present study, obvious chlorosis and lower Fe content were also observed in Cd-treated cucumber leaves. In accordance with the chlorosis, the pigments were significantly reduced and resulted in lower P<sub>N</sub> which led to the inhibition of cucumber plant growth. In *Brassica napus*, *Brassica juncea* and *Lepidium sativum* L. pigment contents, photosynthesis and plant growth were also significantly reduced by Cd treatments (Raziuddin *et al.*, 2011; Gill *et al.*, 2012), and Gill *et al.*, (2012) thought that the physiological disorder greatly depended on nitrogen metabolism perturbation by Cd. Accompanied with the reduction of P<sub>N</sub>, lower G<sub>s</sub> was simultaneously observed in cucumber leaves, nevertheless, the reduction of P<sub>N</sub> did not result from a low stomatal conductance (G<sub>s</sub>) and low CO<sub>2</sub> concentration in chloroplasts, because the intracellular CO<sub>2</sub> concentration (C<sub>i</sub>) levels were much higher in leaves treated with Cd compared to the control (Table 1). Based on this, it could be concluded the inhibition of photosynthetic processes by Cd in cucumber was attributed to nonstomatal restriction (Feng *et al.*, 2010). The Fv/Fm ratio is used to express the photochemical efficiency of PSII, in the present study, Cd treatment caused significant decrease of Fv/Fm, which is a reliable sign of photoinhibition (Krause, 1988), and at the same time, lower Fv/Fm also showed that the ability of PSII reducing the primary acceptor Q<sub>A</sub> was inhibited by Cd (Babani *et al.*, 1996).

Nitric oxide, as a gas signal molecule, can readily form complexes with transition metal ions in aqueous solutions or those present in diverse nucleophilic compounds such as metalloproteins (Stamler *et al.*, 1992). The Fe (III)NO complex appears to undergo a charge transfer reaction to form Fe(II)NO<sup>+</sup> and NO could increase the availability of iron in plants (Graziano *et al.*, 2002). In the present experiment, application of SNP reversed the chlorosis of cucumber leaves induced by Cd stress. As NO donor SNP contains iron in its molecule, it is important to exclude the effect of iron in SNP for the elucidation of NO function. Therefore, potassium

ferrocyanide, an analog of SNP that does not release NO was used to treat cucumber instead. The results showed that it had no significant effect on Fe content and Cd-induced cucumber leaf chlorosis. Therefore, NO from SNP increasing the Fe content might play important role in alleviation of cucumber leaf chlorosis induced by Cd. There were some other investigations indicated that NO was a key component in the regulation of iron uptake and homeostasis in plants (Chen *et al.*, 2010b; Graziano & Lamattina, 2007). In accordance with the chlorosis alleviation by application of SNP, photosynthesis of cucumber was significantly prompted and plant growth was increased, similar function of SNP has been observed in Cd-treated barley (Chen *et al.*, 2010a).

Excessive production of reactive oxygen species (ROS) was found in several plant treated with Cd (Markovska *et al.*, 2009; Chen *et al.*, 2010a; Dai *et al.*, 2012). In the experiment, Cd treatment dramatically induced accumulation of H<sub>2</sub>O<sub>2</sub>. A direct result of excessive ROS is lipid peroxidation which is often indicated as TBARS contents, and the accumulation of TBARS was considered to be caused by oxidative degradation of polyunsaturated fatty acids, in particular linolenic acid which is localized mainly in the thylakoid glycolipids, therefore, excessive TBARS accumulation in Cd treated cucumber leaves is likely a good measure for peroxidative damage to chloroplast membrane (Van-Hasselt *et al.*, 1996).

It is well known higher efficacy of antioxidant system plays an important role in keeping ROS balance in plant tissues under stress conditions (Kafi *et al.*, 2011). In the present study, we found that Cd dramatically inhibited activities of antioxidant enzymes and lowered GSH contents, which could mainly be responsible for oxidative stress induced by Cd. Above results indicated that exogenous SNP could reduce Cd-induced oxidative stress. To confirm the possible mechanism, antioxidant capacities were determined in the present study. It was observed that exogenous SNP significantly reversed the inhibition of all antioxidant capacities indicated as DPPH-radical scavenging activity, hydroxyl radical scavenging activity and the chelating activity in cucumber leaves under Cd stress. Increase of antioxidant capacity in plants usually depends on antioxidant metabolites such as glutathione, polyphenols and flavonoids (Ksouri *et al.*, 2007), and

there were reports indicating that exogenous NO could induce synthesis of these antioxidant metabolites (Ksouri *et al.*, 2007; Ferreira *et al.*, 2010; Wu *et al.*, 2007), especially glutathione, it could be increased by NO modulating both synthesis pathway (Innocenti *et al.*, 2007; Moellering *et al.*, 1998) and reduction pathway (Xu *et al.*, 2010). Glutathione (GSH) content in cucumber leaves was increased by exogenous SNP treatment under Cd stress condition, this should be one important reason that exogenous SNP increased antioxidant capacity of cucumber leaves under Cd stress. Except for antioxidant substances, several ROS-scavenging enzymes including SOD, CAT, GPX, APX, DR and GR play important role in alleviating oxidative stress (Nabati *et al.*, 2011). Cheng *et al.*, (2002) reported that the inhibition of polyethylene glycol (PEG)- and dehydration (DH)-enhanced senescence of rice leaves by NO was most likely modulated through increasing SOD activity which resulted in lower lipid peroxidation. In the present experiment, exogenous SNP significantly alleviated the inhibited level of SOD activity by Cd stress (Table 2), which suggested that application of SNP could promote the conversion from  $O_2^-$  to  $H_2O_2$  and  $O_2$  in cucumber leaves under Cd stress. It is well known that SOD includes Mn-SOD, Cu,Zn-SOD and Fe-SOD, and it could be able to conclude that the increasing SOD activity by exogenous SNP might be partly attributed to higher Fe efficiency.  $H_2O_2$ , as the products of  $O_2^-$  decomposition, can rapidly diffuse across the membrane and is toxic because it acts both as an oxidant as well as reductant (Foyer, 1994), under such situation, the highly efficiency of  $H_2O_2$  scavenging is very vital. In our experiment, exogenous SNP significantly increased CAT and GPX activities (Table 2), which can directly scavenge  $H_2O_2$  and play important roles in reducing oxidative stress induced by abiotic stress, similar results were obtained in heat stressed chrysanthemum (Yang *et al.*, 2011). Besides SOD, CAT and GPX, there exists another important ROS-scavenging system which is ascorbate-glutathione cycle in chloroplast (Murshed *et al.*, 2008). The cycle is mainly composed of enzymes such as APX, DR, GR and antioxidants such as glutathione and ascorbate. In our experiment, exogenous SNP significantly induced activities of APX, DHAR and GR (Table 2), which are very important in protecting chloroplast from oxidative damage and might be greatly responsible for the higher photosynthesis capacity of Cd-treated cucumber in the present study.

## Conclusion

Exogenous SNP application effectively reversed the chlorosis and alleviating growth inhibition of cucumber induced by Cd stress, while application of potassium ferrocyanide, an analog of SNP that does not release NO, did not showed the effects. Based on the results, it could conclude that the effects of SNP alleviating Cd stress on cucumber was due to NO from it, and the mechanism might depend on higher Fe efficiency and antioxidant efficiency induced by SNP.

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